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**Computation Notebook**

113/4" x 91/4", 4 x 4 Quad., 75 Sheets

**43-648**



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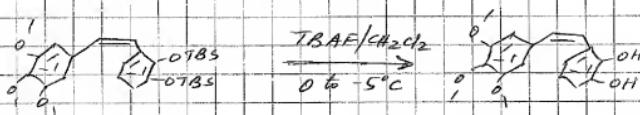


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## DEPROTECTION USING TBAF

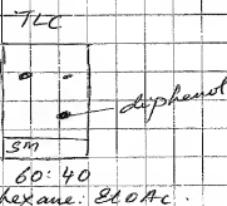
07/04/02



MF	SM	TBAF in 1M THF	CH <sub>2</sub> Cl <sub>2</sub> dry Solvent
MW	530.84	261.47	
Ratio	1	2.5	
wt(g)	2.02	2.49 (9.53ml)	
mmoles	3.81	9.53	

## PROCEDURE:

1. Dried the compound on pump for  $\frac{1}{2}$  hour.
2. Put under Ar.
3. Dissolve in CH<sub>2</sub>Cl<sub>2</sub>.
4. Maintain temperature at -5°C by using ice bath to which NaCl is added.
5. TLC was done after 15 min, 20, 25, 30, 35 min.
6. Reaction was stopped after 35 min by quenching with H<sub>2</sub>O.



7. Transferred to a separatory funnel and sonicated more H<sub>2</sub>O and do partitioning.

- 8 Collect the lower organic layer
- 9 Extract the aqueous layer with  $\text{CH}_2\text{Cl}_2$  twice
- 10 The combined organic layers were dried with anhydrous  $\text{Na}_2\text{SO}_4$
- 11 Filtered & rotary evaporated

### COLUMN CHROMATOGRAPHY.

07/05/02

Did a column to collect purified diol.

Used 70:30 hexane:  $\text{Et}_2\text{OAc}$

wt of diol collected = 0.5g.

### NMR

$^1\text{H}$  NMR PA - 11 - 56

$^{13}\text{C}$  NMR PA - 11 - 57

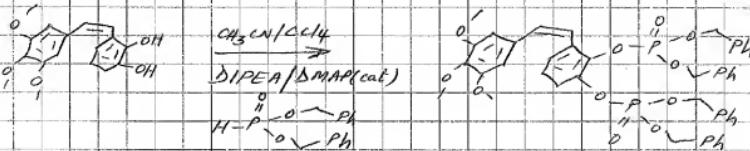


$$\text{yield \%} = \frac{0.5}{302.32} = 1.66 \text{ mmoles}$$

$$\frac{1.66}{3.81} \times 100 = 43.40\%$$

## DIBENZYL PHOSPHORYLATION OF 3,04

07/06/02

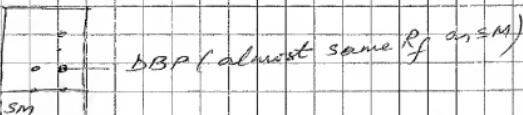


MF	3M	CCl <sub>4</sub>	DIPEA	DMAP	DBP	KH <sub>2</sub> PO <sub>4</sub> (0.5M)	CH <sub>3</sub> CN
MW	302	153.82	129.24	122.17	262.25	136.09	
wt(g)	0.5(g)	2.53(16ml)	0.89(1.2ml)	0.04	1.26(1.06)		10ml
Ratio	1eq	10eq	4.2eq	0.2eq	2.7eq		
mmoles	1.65	16.5	6.95	0.33	4.8		

## PROCEDURE Ref Petit Antineoplastic agents 42.9 pg 207

1. The diphenol was put Ar, cool to -20°C
2. Add CH<sub>3</sub>CN, then add CCl<sub>4</sub> and stir for 10 min
3. Diisopropylethyl amine and DMAP were then added
4. 1 minute later dibenzyl phosphate was added and temperature was maintained below -20°C
5. After 4.5 min do TLC to check compound formation

TLC

60:40  
hexane EtOAc

6. Then KH<sub>2</sub>PO<sub>4</sub> (0.5M) was added and the

7. It was extracted with Et<sub>2</sub>OAc, 4 times then washed with saturated NaCl and water and dried with anhydrous NaCl

8. It was rotary evaporated and the yellowish brown mixture was separated by column chromatography.

#### COLUMN CHROMATOGRAPHY

Separated by flash chromatography using 70:30 hexane : Et<sub>2</sub>OAc

Then use 60:40 hexane : Et<sub>2</sub>OAc

Colorless oily liquid wt = 0.71 g.

<sup>1</sup>H NMR PA - II - 58

<sup>31</sup>P NMR PA - II - 59

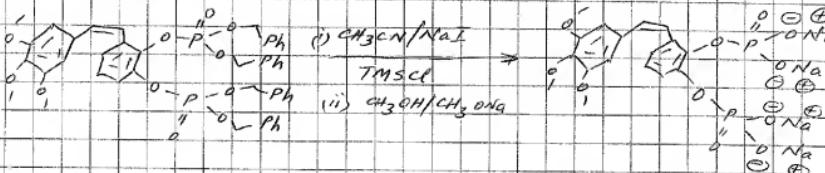
<sup>13</sup>C NMR PA - II - 60

$$\text{yield} = \frac{0.71}{822.77} = .86 \text{ mmoles}$$

$$\% = \frac{0.86}{1.65} \times 100 = 52.29\%$$

## TETRASODIUM - DIPHOSPHATE

07/09/02



MF	SM	NaI	TMSCl	$\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$	$\text{CH}_3\text{ONa}$ 25%	$\text{CH}_3\text{CN}$	$\text{CH}_3\text{OH}$
		832					
MW	822	149.89	108.64	248	54.02	9.5ml	
WT(g)	0.71	0.52	0.37 (0.45ml)	1ml	0.19 (0.8ml)	2.8	
mmoles	0.86	3.44	3.44			3.44	
Ratio	1eq	4eq	4eq	1%		4eq	

PROCEDURE: Ref Petit Antineoplastic agents 429 pg 207.

1. The phosphate was out under Ar.
2.  $\text{CH}_3\text{CN}$  was added to it followed by  $\text{NaI}$ .
3. The mixture was stirred for 2 mins, then chlorotrimethylsilane (distilled using  $\text{CaH}_2$ ) was added dropwise.
4. 30 min later TLC was done to ensure that the SM was used up.

TLC

60:40 hexane :  $\text{EtOAc}$ .

5. Reaction was terminated by adding 1%  $\text{Na}_2\text{S}_2\text{O}_3$ .

7. The mixture was dissolved in  $\text{H}_2\text{O} - \text{CH}_2\text{Cl}_2$
8. The  $\text{CH}_2\text{Cl}_2$  layer was washed with  $\text{H}_2\text{O}$  three times
9. The aqueous layer was concentrated using toluene azeotrope
10. The residue was dried on vacuum pump overnight
11. The dry mixture (looked like foam and was dark brownish in colour) was dissolved in  $\text{CH}_3\text{OH}$  then  $\text{CH}_3\text{ONa}$  was added and the solution was stirred for 8 hours

## RECRYSTALLISATION

07/12/02

Tried recrystallising using  $\text{H}_2\text{O}$  - Ethanol  
 $\text{H}_2\text{O}$  - methanol  
 $\text{H}_2\text{O}$  - acetone

did not work.

Separated 40mg using prep TLC (70 : 30 water : isopropanol)

Got 10mg of the trans isomer

Separated 15 fractions by prep TLC

SEPARATION USING C-18 COLUMN

07/15/02

Tried separation using  $\text{CH}_3\text{OH} : \text{H}_2\text{O} : \text{CH}_3\text{CN}$  system? R<sub>trans</sub> 0.35 : 1.0 : 3.5

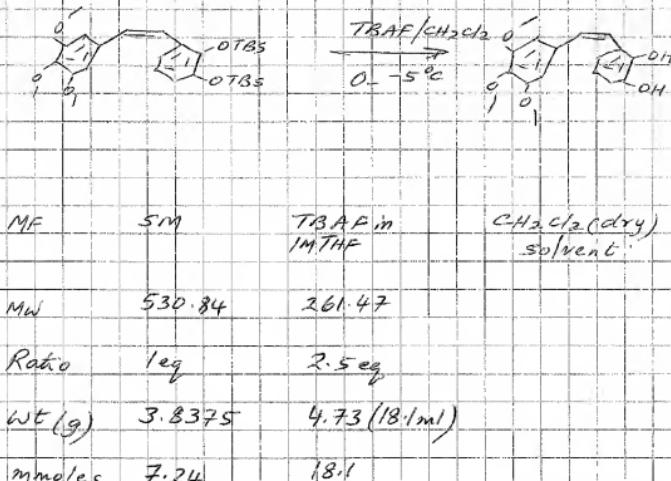
using C18 column first run methanol towards it (least polar)

The compound came out as a mixture.

Tried separation on prep TLC. It was a disaster

## DEPROTECTION USING TBAF

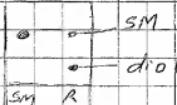
07/30/01



## PROCEDURE:

1. The compound was put under Ar and dissolved in  $\text{CH}_2\text{Cl}_2$ .
2. It was cooled to  $-5^\circ\text{C}$  by adding common salt to ice bath.
3. TBAF was added and TLC done after 1.5 mins.

## TLC



70:30 hexane:2tBAc

4. Reaction was continued for 50 min till all the SM was used up.

TLC



70:30 hexane: 2tOAc

5. The reaction was quenched with water & extracted the aqueous layer with  $\text{CH}_2\text{Cl}_2$  twice.
6. The combined organic layers were dried over  $\text{Na}_2\text{SO}_4$  anhydrous, filtered and concentrated.
7. The dark brown colour oily crude was purified by column chromatography.

#### Column CHROMATOGRAPHY.

The compound was purified by flash column chromatography using hexane: 2tOAc 60:40 to yield 2.05g of product.

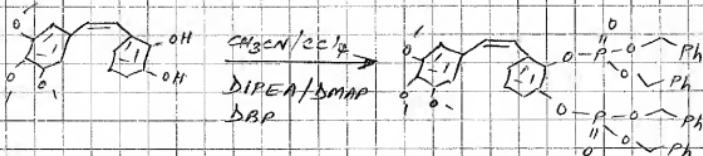
$^1\text{H-NMR}$  PA-11-81

$$\text{yield \%} = \frac{2.05}{302.32} = 6.78 \text{ mmoles}$$

$$\frac{6.78 \times 100}{7.24} = 93.65\%$$

## BIBENZYL PHOSPHORYLATION

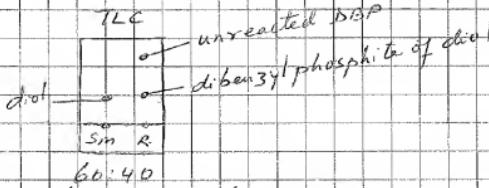
08/01/02



MF	SM	CCl <sub>4</sub>	DIPEA	DMAP	DBP	CH <sub>3</sub> CN	KH <sub>2</sub> PO <sub>4</sub> 0.5M
302.32							
MW	886	153.82	129.24	122.17	262.25	136.09	
wt(g)	0.86	4.46 (8ml)	1.57 (2.2ml)	0.07	2.2 (1.85ml)	17ml	
Ratio	1eq	10eq	4.2eq	0.2eq	2.9eq		
mmoles	2.85	28.5	12.2	0.57	8.3		

PROCEDURE - Ref GR Palkar, Anti-Cancer Drug Design (2000),  
15, 203-216

1. The diphenol was put under Ar.
2. It was cooled to -20°C. Then CCl<sub>4</sub> was added and stirred for 10 min.
3. DIPEA and DMAP were then added.
4. After stirring for 1 min, dibenzylphosphate was added and the temperature maintained below -20°C.
5. After 45 min, TLC was performed.



6. The *o*-benzylphosphorylated compound appears at almost the same *Rf* as diol (watch carefully).
7. The reaction was quenched with 0.5 M  $\text{KH}_2\text{PO}_4$  and the mixture was allowed to come to r.t.
8. The aqueous layer was extracted with Et<sub>2</sub>O four times, washed with brine, then with  $\text{H}_2\text{O}$  and dried using  $\text{Na}_2\text{SO}_4$  anhydrous.
9. It was filtered and concentrated.

## COLUMN CHROMATOGRAPHY

08/02/02

The compound was loaded on column (wet loading) and flash column chromatography was performed using 3:2 hexane-ethyl acetate to afford 1.8 g of a colourless oil.

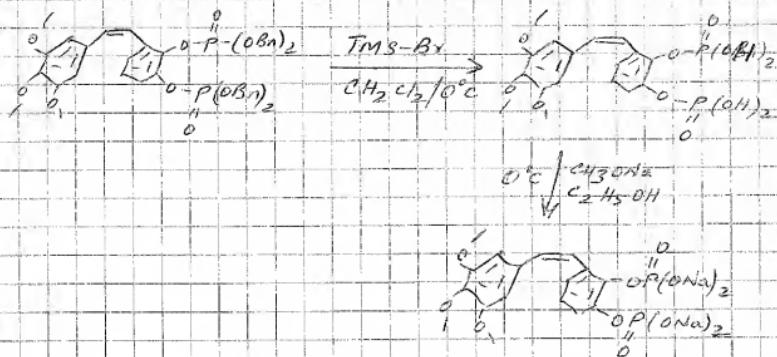
<sup>1</sup>H NMR PA-11-82<sup>31</sup>P-NMR PA-11-83

$$\text{yield \%} = \frac{1.73}{8.22-7.7} = 2.10 \text{ mmoles}$$

$$\frac{2.10}{2.85} = 73.77\%$$

## TETRA-SODIUM Di-PHOSPHOMATE

08/03/202



PROCEDURE: Ref C. R. Peltier et al. Ant. Cancer Drug Design (2001)  
16, 185-193

MF	SM	TMSBr	CH <sub>2</sub> Cl <sub>2</sub>	C <sub>2</sub> H <sub>5</sub> OH	CH <sub>3</sub> ONa
MW	822	153.10			54.02
Wt(g)	0.12	0.089 (0.08ml)	1.1ml	2.2ml	0.032
mmoles	0.146	0.583			0.583
Ratio	1eq	4eq			4eq

Na<sub>2</sub>SO<sub>3</sub> 0% → 1:1 m/l

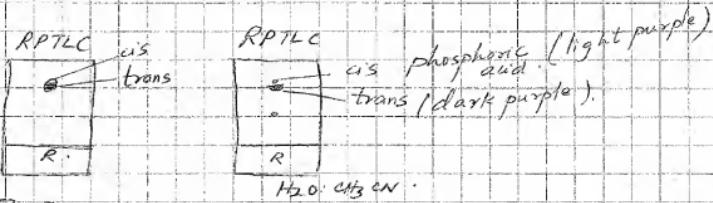
## PROCEDURE:

1. The tetra-benzylic ester was put under Ar.
2. It was dissolved in DCM at 0°C & bromo tri-n-butyl silane was added.
3. After stirring for 15 min, TLC was done.

TLC

Front quickly  
+ disappear (diben3yl bromide)

4. The reaction was quenched with 10%  $\text{Na}_2\text{S}_2\text{O}_3$  (aq.) after 25 min.



water: isopropanol

5. Isomerization had already taken place with a greater percentage to going to trans and there was minor formation of cis.

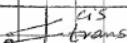
6. The solvent aqueous layer was removed with a disposable pipette. It was extracted with  $\text{EtOAc}$  three. The combined organic layers were rotavaped and then dried on the vacuum pump. It took very long to dry even after 12 hours, it wasn't looking very dry.

7. It was then put under  $\text{Ar}$ , maintained at  $0^\circ\text{C}$  and dissolved in  $\text{C}_2\text{H}_5\text{OH}$ .

8. Sodium methoxide was quickly added and stirred for 30 min. Light yellow colour precipitate was obtained which was rotavaped.

9. It was titrated with ether, turned dark brown. TLC showed greater formation of trans.

RPTLC

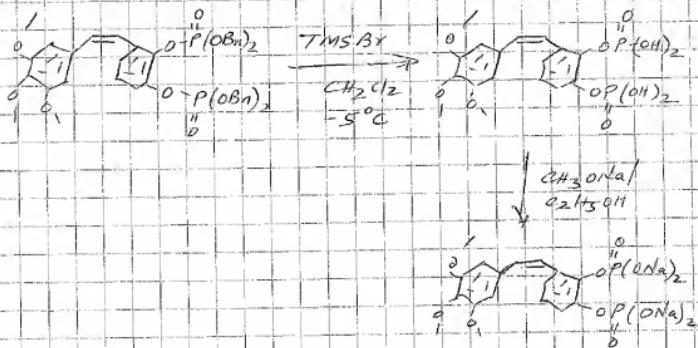


70:30 water: isopropanol

10. RECRYSTALLIZATION from water, acetone was attempted but failed.

## TETRA SODIUM DIPHOSPHATE

08/05/02



PROCEDURE: Ref G.R.Petit et al. *Anti Cancer Drug Design* (2001)  
16 pg 185 - 193

SMF	SM	TMSBr	CH <sub>2</sub> Cl <sub>2</sub>	CH <sub>3</sub> ONa	C <sub>2</sub> H <sub>5</sub> OH	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>
ML	822	153.10			54.02	
WT (g)	0.100	0.075 (0.044) 0.9 ml		0.026	1.8 ml	0.9 ml
mmoles	0.12	0.486		0.486		
Ratio	1 eq	4 eq		4 eq		

PROCEDURE: Ref G.R.Petit et al. *Anti Cancer Drug Design* (2001)  
16, 185 - 193

1. To a stirred solution of tetra-benzyl ester in CH<sub>2</sub>Cl<sub>2</sub> under Ar at 0°C (-5°C), bromotrimethylsilane was added.
2. After stirring for 5 min, TLC was done.

TLC

o	faint and quickly disappeared
o	perhaps dibenzyl bromide
sm R	is highly volatile

60:40

hexane:EtOAc

RP TLC

o	
	R

H<sub>2</sub>O:CH<sub>3</sub>CN  
70:30

RPTLC

o	
	R

H<sub>2</sub>O:Isopropanol  
70:30phosphoric  
acid water  
medium

3. However the reaction was quenched with 1% Na<sub>2</sub>S<sub>2</sub>O<sub>8</sub> in CH<sub>3</sub>CO<sub>2</sub>H (Note no water was used) only after 15 min on the RP TLC plates took long to develop.

4. The solution was rotavaped using Toluene azeotrope & then CH<sub>2</sub>Cl<sub>2</sub>.

5. A TLC was performed after rotavaping off the solvent

RP TLC

o	cis (light purple)
o	trans/dark purple
=	
sm	

70:30 Water: Isopropanol

6. Perhaps the isomerization took place after 5 min

7. The compound (crude) was dried light yellow mixed with brown (mustard colour) on the vacuum pump for 6 hours. It dried and was like a foam

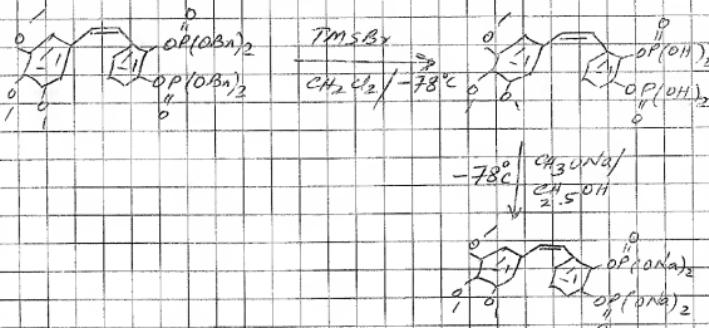
8. The dried crude was put under Ar, dissolved in CsH<sub>5</sub>OH at 0°C.

9. NaOCH<sub>3</sub> was quickly added and stirred for 30 mins. The compound was taken out, it was white in colour and rotavaped.

10. It was triturated with anhydrous diethyl ether

## TETRA SODIUM DI PHOSPHATE

08/07/02



MF	SM	TMSCl	CH <sub>2</sub> Cl <sub>2</sub>	CH <sub>3</sub> ONa	C <sub>2</sub> H <sub>5</sub> OH	Na <sub>2</sub> PO <sub>4</sub> (aq)
MW	822	153.10			54.02	
wt(g)	0.13	0.098	1.2ml	0.035	2.4ml	1.2ml
mmoles	0.16	0.64			0.64	
Ratio	1 eq	4 eq			4 eq	

PROCEDURE Ref GR Petit et al, Antineoplastic agents 460  
*Anti Cancer Drug Design (2001) 16, 185-193*

1. The tetra-benzyl ester was put under Ar
2. Temp was -78°C. It was dissolved in dichloro-methane
3. TMSCl was added & temp maintained at -78°C
4. After stirring for 5 min, 15 min, 30 min, 45 min. TLC was done.

TLC (5 min)

very light

TLC (15 min)

a faint

o o

5m R

60:40

hexane:EtOAc

TLC (30 min)

Darker  
benzyl bromide

o o faint

5m

60:40

hexane:EtOAc

TLC (45 min)

a darker

o

5m

60:40

hexane:EtOAc

5. After stirring for 4.5 min the rxn was quenched with 10%  $\text{Na}_2\text{S}_2\text{O}_3$  (aq) and stirred for 5 more min. It looked like a creamy ppt mixed with an oily layer.

6. The creamy layer was removed using a disposable pipette into a t.t. It was extracted with EtOAc twice.

7. The combined organic layers were rotavaped. Finally a little NaCl was added and rotavaped.

8. A colourless oily liquid (phosphinic acid) intermediate was obtained which was dried on the vacuum pump.

RP TLC

✓ — dark grey  
 —? purple

60:30  
 water:  $\text{CH}_2\text{Cl}_2$